

ANTITUMOR AGENTS, 74.<sup>1</sup> BRUCEANOL-A AND -B, TWO NEW ANTILEUKEMIC QUASSINOIDS FROM *BRUCEA ANTIDYSENTERICA*

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Eight quassinoids, including the antileukemic compound bruceantin, which was in phase II clinical trial by the National Cancer Institute, have been isolated from the Ethiopian tree *Brucea antidysenterica* Mill. (Simaroubaceae) by Kupchan and associates (1). In a previous paper, we reported the isolation and structural elucidation of two new antileukemic quassinoid glycosides, bruceantinoside-A and -B, from *B. antidysenterica* (2). We now describe the isolation and characterization of two additional new antileukemic isobruceine type quassinoids, bruceanol-A (**1**) and -B (**2**), from this same plant.

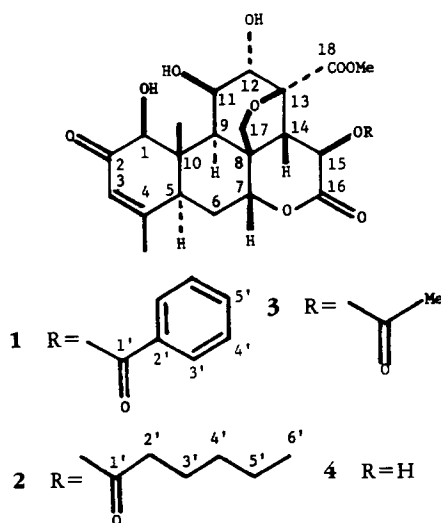
## RESULTS AND DISCUSSION

Bruceanol-A (**1**) had the molecular formula  $C_{28}H_{30}O_{11}$  from both field de-

sorption and hrms. Compound **1** showed the presence of hydroxyl ( $3540$ ,  $3450$ , and  $1030\text{ cm}^{-1}$ ), phenyl ( $1600$  and  $1580\text{ cm}^{-1}$ ),  $\delta$ -lactone ( $1750\text{ cm}^{-1}$ ), ester ( $1730\text{ cm}^{-1}$ ), and  $\alpha$ ,  $\beta$ -unsaturated carbonyl ( $1675\text{ cm}^{-1}$ ) groups in its ir spectrum. The  $^1\text{H}$ -nmr spectrum of **1** is identical with that of isobruceine-B (**3**), except for the signals in the acid portion of the ester as well as the COOMe-13 and the H-15. These differences were observed in **1** as a phenyl ring [ $\delta$  8.0-8.05 (2H), H-3' and 7.42-7.65 (3H), H-4' and H-5'] and in **3** as a methyl group [ $\delta$  2.08 (3H, s)]. In **1**, the signal for COOMe-13 ( $\delta$  3.51) was shifted upfield by 0.32 ppm, and that for H-15 ( $\delta$  6.49) was shifted downfield by 0.19 ppm compared to the corresponding signals of **3** ( $\delta$  3.83 for COOMe-13 and  $\delta$  6.30 for H-15) due to the anisotropic effect of the aromatic ring in **1** (Table 1).

Added confirmation for these assignments was obtained by a comparison of the  $^{13}\text{C}$ -nmr spectra of **1** and **3** (Table 2). The carbon signals for both compounds were superimposable except for the differences in the above mentioned ester side chains. Thus, **1** contained the aromatic carbons at  $\delta$  128.7 (s, C-2'), 128.5 (d, C-3'), 130.0 (d, C-4'), and 133.8 (d, C-5'), whereas **3** possessed a quartet methyl group at  $\delta$  20.5 (C-2').

Bruceanol-B (**2**) was analyzed for  $C_{27}H_{36}O_{11}$ . Its ir spectrum was consistent with the presence of hydroxyl ( $3540$ ,  $3450$ , and  $1030\text{ cm}^{-1}$ ),  $\delta$ -lactone ( $1750\text{ cm}^{-1}$ ), ester ( $1730\text{ cm}^{-1}$ ), and  $\alpha$ ,  $\beta$ -unsaturated carbonyl ( $1677\text{ cm}^{-1}$ ) groups. Compound **2** showed



<sup>1</sup>For part 73, see H. Nozaki, H. Suzuki, T. Hirayama, R. Kasai, R.Y. Wu, and K.H. Lee, *Phytochemistry*, (in press).

also comparable  $^1\text{H}$ -nmr (Table 1) and  $^{13}\text{C}$ -nmr (Table 2) spectra to those of **3** except for the signals attributable to a hexanoyl group [ $\delta$  1.3 (m, H-3', H-4', and H-5') and 0.91 (t,  $J=7.0$  Hz, H-6'), and  $\delta$  33.6 (t, C-2'), 31.2 (t, C-3'), 24.2 (t, C-4'), 22.2 (t, C-5'), and 13.8 (q, C-6')] in **2** and an acetyl [ $\delta$  2.08 (s) and  $\delta$  20.5 (q, C-2')] moiety in **3**.

Deacetylisoruceine-B (**4**) was obtained by alkaline hydrolysis of **1, 2**, and **3**. Thus, **1** and **2** were structurally identical with **3** except for the slight differences in the C-15 ester side chain, in which **1** and **2** have a benzoate and an hexanoate, respectively, whereas **3** bears an acetate moiety.

an internal standard. Mass spectra were recorded on a JEOL JMS-D100 (EI) and/or a Hitachi M-80 (EI and FD) instruments. Silica gel (Merck, type 60, 70-230 mesh) was used for column chromatography, and precoated silica gel plates (Merck, 60 F-254, 0.25 and 2mm) were used for analytical and preparative tlc, respectively. Detection of components was made by use of a uv lamp. Hplc was performed on a Waters Associates Model ALC/GPC 244 liquid chromatograph using a  $\mu$ Bondapak C18 column (3.9 mm $\times$ 300 mm) for analytical, and M and S PAK C18-A (20 mm $\times$ 250 mm) for preparative purposes.

CHROMATOGRAPHY OF THE  $\text{CHCl}_3$  FRACTIONS.—The crude  $\text{CHCl}_3$  fraction (372 g), which is part of the  $\text{CHCl}_3$  extract of the ground wood of *B. antidyenterica* (4228 lbs) reported previously (2), was subjected to column chromatography on silica gel (2 kg, 10 cm $\times$ 60 cm and

TABLE 1.  $^1\text{H}$ -nmr Spectra of Quassinoids **1, 2, 3**, and **4**

	Bruceanol-A ( <b>1</b> ) <sup>a</sup>	Bruceanol-B ( <b>2</b> ) <sup>b</sup>	Isobruceine-B ( <b>3</b> ) <sup>c</sup>	Deacetylisoruceine-B ( <b>4</b> ) <sup>d</sup>
H-1	4.32 s	4.28 br.s	4.26 s	4.85 br.s
H-3	6.13 br.s	6.1 br.s	6.11 br.s	6.13 br.s
H-7	4.8 m	4.75 br.s	4.75 m	5.54 t (4.9) <sup>e</sup>
H-11	4.57 s	4.54 br.s	4.75 m	5.08 br.s
H-12	4.20 s	4.18 br.s	4.12 br.s	4.35 br.s
H-15	6.49 d (13) <sup>e</sup>	6.29 d (13) <sup>e</sup>	6.30 d (13) <sup>e</sup>	6.12 d (12.2) <sup>e</sup>
Ome	3.51 s	3.83 s	3.83 s	3.79 s
Me-4	1.98 s	1.96 s	1.95 s	1.72 s
Me-10	1.21 s	1.19 s	1.20 s	1.44 s
2'	—	—	2.08 s	—
3'	—	—	—	—
4'	8.0-8.05 (2H)	1.3 m	—	—
5'	7.42-7.65 (3H)	—	—	—
6'	—	0.91 t (7) <sup>e</sup>	—	—

<sup>a</sup>Measured at 200 MHz in  $\text{CDCl}_3$ .

<sup>b</sup>Measured at 400 MHz in  $\text{CDCl}_3$ .

<sup>c</sup>Measured at 400 MHz in  $\text{CDCl}_3$  by Polonsky *et al.* (3).

<sup>d</sup>Measured at 400 MHz in  $\text{C}_5\text{D}_5\text{N}$ .

<sup>e</sup>Coupling constant in Hz.

## EXPERIMENTAL

### GENERAL EXPERIMENTAL PROCEDURES.—

Melting points were determined on an MRK air-bath-type melting point apparatus and were uncorrected. Specific rotations were obtained on a Yanagimoto OR-50 polarimeter (L=0.5 dm). Ir and uv spectra were recorded on a Hitachi 215 grating ir spectrometer and a Hitachi 320-S uv spectrometer, respectively.  $^1\text{H}$ -nmr and  $^{13}\text{C}$ -nmr spectra were determined on a JEOL JNM-FX 200 (200 MHz for  $^1\text{H}$  nmr and 50.1 MHz for  $^{13}\text{C}$  nmr) and/or a JEOL JNM-GX400 (400 MHz for  $^1\text{H}$  nmr and 100 MHz for  $^{13}\text{C}$  nmr) using TMS as

eluted first with  $\text{CHCl}_3$  and then with increasing amounts of MeOH in  $\text{CHCl}_3$  to yield 28 fractions. Fraction 20 (31 g) was further subjected to column chromatography on silica gel (500 g, 4 cm $\times$ 90 cm and eluted with  $\text{EtOAc-Et}_2\text{O}$ , 1:1, v/v) to afford a pale yellow gum (6.7 g) which showed the presence of isobruceine-B (**3**), bruceanol-A (**1**), and bruceanol-B (**2**) on analytical hplc (MeOH- $\text{H}_2\text{O}$ , 1:1, v/v).

ISOLATION OF BRUCEANOL-A (**1**).—The pale yellow gum (6.7 g) was subjected to preparative hplc using a mixed solvent of MeOH- $\text{H}_2\text{O}$  (1:1, v/v) to give a colorless amorphous compound (1.1

TABLE 2.  $^{13}\text{C}$ -nmr Spectra of Quassinoids **1**, **2**, **3**, and **4**

	Bruceanol-A ( <b>1</b> ) <sup>a</sup>	Bruceanol-B ( <b>2</b> ) <sup>b,c</sup>	Isobruceine-B ( <b>3</b> ) <sup>d</sup>	Deacetylisobruceine-B ( <b>4</b> ) <sup>e</sup>
C-1	80.7 d	80.6 d	81.3 d	82.9 d
C-2	196.7 s	196.9 s	197.6 s	198.4 s
C-3	124.3 d	124.3 d	124.5 d	125.0 d
C-4	163.0 s	162.9 s	162.6 s	163.2 s
C-5	43.5 d	43.5 d	43.4 d	43.9 d
C-6	28.6 t	28.5 t	28.2 d	28.6 t
C-7	81.3 d	81.1 d	81.7 d	83.1 d
C-8	45.8 s	45.7 s	45.8 s	46.2 s
C-9	42.8 s	42.7 d	42.4 d	43.4 d
C-10	47.6 s	47.5 s	47.7 s	48.4 s
C-11	72.7 d	72.5 d	74.3 d	75.2 d
C-12	75.8 d	75.8 d	75.1 d	76.9 d
C-13	82.9 s	82.8 s	81.7 s	82.6 s
C-14	51.4 d	51.5 d	52.3 d	55.5 d
C-15	67.4 d	66.5 d	67.8 d	66.3 d
C-16	166.7 s	166.9 s	167.6 s	172.8 s
C-17	73.3 t	73.3 t	73.0 t	74.0 t
C-18	172.0 s	172.4 s	169.5 s	173.5 s
OMe	52.9 q	52.8 q	49.8 q	52.3 q
Me-4	22.5 q	22.5 q	22.4 q	22.2 q
Me-10	11.6 q	11.5 q	11.3 q	11.4 q
C-1'	164.8 s	171.9 s	170.7 s	—
C-2'	128.7 s	33.6 t	20.5 q	—
C-3'	128.5 d	31.2 t	—	—
C-4'	130.0 d	24.2 t	—	—
C-5'	133.8 d	22.2 t	—	—
C-6'	—	13.8 q	—	—

<sup>a</sup>Measured at 50.1 MHz in  $\text{CDCl}_3$ .<sup>b</sup>Measured at 100.40 MHz in  $\text{CDCl}_3$ .<sup>c</sup>The characterizations of sec-C was based on the INEPT (Insensitive Nuclei Enhanced by Polarization Transfer).<sup>d</sup>Measured at 22.63 MHz in  $\text{CDCl}_3$  by Polonsky *et al.* (3).<sup>e</sup>Measured at 100.40 MHz in  $\text{C}_5\text{D}_5\text{N}$ .

g), which was eluted after isobruceine-B. This amorphous compound was further subjected to preparative hplc to yield pure bruceanol-A (**1**, colorless and amorphous, 0.000058% yield); mp 174–177° (from  $\text{CH}_2\text{Cl}_2/\text{Et}_2\text{O}$ );  $[\alpha]^{25\text{D}} + 60^\circ$  (c 0.5, EtOH); uv  $\lambda$  max (EtOH) 232 nm ( $\epsilon$  22400); The ir spectral data were discussed in the text. The  $^1\text{H}$ -nmr and  $^{13}\text{C}$ -nmr data were summarized in Table 1 and 2, respectively. Fdms  $m/z$  543 ( $\text{M}^+ + 1$ ); eims  $m/z$  542.1853 (calcd. for  $\text{C}_{28}\text{H}_{30}\text{O}_{11}$ : 542.1788,  $\text{M}^+$ ), 524, 1657 (calcd. for  $\text{C}_{28}\text{H}_{28}\text{O}_{10}$ : 524.1682,  $\text{M}^+ - \text{H}_2\text{O}$ ), 492, 1418 (calcd. for  $\text{C}_{27}\text{H}_{24}\text{O}_9$ : 492.1420,  $\text{M}^+ - \text{MeOH}$ ), and 420, 1438 (calcd. for  $\text{C}_{21}\text{H}_{24}\text{O}_9$ : 420.1420,  $\text{M}^+ - \text{C}_6\text{H}_5\text{COOH}$ ).

**ISOLATION OF BRUCEANOL-B (2).**—The pale yellow gum (6.7 g) was further eluted after the elution of bruceanol-A to afford a compound (0.54 g). This compound was purified by preparative hplc to yield pure bruceanol-B (**2**, colorless and amorphous, 0.000028% yield); mp 170–172° (from EtOAc/ $\text{Et}_2\text{O}$ );  $[\alpha]^{25\text{D}} + 32^\circ$  (c 0.5,

MeOH); uv  $\lambda$  max (EtOH) 236 nm ( $\epsilon$  8300); fdms  $m/z$  537 ( $\text{M}^+ + 1$ ); Found: C, 60.49%; H, 6.74%. Calcd. for  $\text{C}_{27}\text{H}_{36}\text{O}_{11}$ : C, 60.43%, H, 6.76%.

**ISOLATION OF ISOBRUCEINE-B (3).**—Isobruceine-B (**3**) was obtained as a colorless and amorphous compound (0.1 g) by the preparative hplc of the foregoing pale yellow gum (6.7 g). It was also obtained in 4% yield from fraction 21 (88 g) in the same manner as described above. This compound was identified by comparing its  $^1\text{H}$ -nmr (400 MHz) spectrum (Table 1) with that presented by Polonsky (3).

**HYDROLYSIS OF ISOBRUCEINE-B (3).**—To a solution of isobruceine-B (**3**, 25 mg) in MeOH (1.5 ml) was added 1N NaOH (0.45 ml). After stirring at  $-15^\circ$  for 17 min, the reaction mixture was neutralized with 1N HCl (0.8 ml) and evaporated in vacuo to give a residue. The residue was subjected to preparative tlc to afford pure deacetylisobruceine-B (**4**) as colorless needles (8.2

mg); mp 279-282° (from EtOAc/Et<sub>2</sub>O); ir (KBr) 3400, 1735 and 1670 cm<sup>-1</sup>; eims *m/z* 438.1538 (M<sup>+</sup>; calcd. for C<sub>21</sub>H<sub>26</sub>O<sub>10</sub>: 438.1526); Found: C, 57.49%; H, 5.99%. Calcd. for C<sub>21</sub>H<sub>26</sub>O<sub>10</sub>: C, 57.53%; H, 5.98%.

**HYDROLYSIS OF BRUCEANOL-A (1).**—Bruceanol-A (1, 46.6 mg) was hydrolyzed and worked up in the same manner described above to give colorless needles (6.2 mg) identified as deacetylisorbruceine-B (4) by comparing its <sup>1</sup>H-nmr (400 MHz, C<sub>5</sub>D<sub>5</sub>N) spectrum with that of an authentic sample derived from the hydrolyzed product of isobruceine-B.

**HYDROLYSIS OF BRUCEANOL-B (2).**—Bruceanol-B (2, 25 mg) was hydrolyzed and worked up in the same manner as described above to yield colorless needles (7.2 mg) identified as deacetylisorbruceine-B (4) by a direct <sup>1</sup>H-nmr comparison with an authentic sample.

**BIOLOGICAL ACTIVITY.**—In vivo antitumor assay was carried out according to a standard National Cancer Institute procedure described in the literature (4). Bruceanol-A (1) and -B (2) showed significant (T/C ≥ 120%) antileukemic activity in vivo against P-388 lymphocytic leukemia (3-day dosing) at T/C = 130% (0.5 mg/kg), 129% (1 mg/kg) and 134% (2 mg/kg), and 123% (0.5 mg/kg), respectively. The control, 5-fluorouracil, had T/C = 135% (200 mg/kg, 1-day dosing).

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